

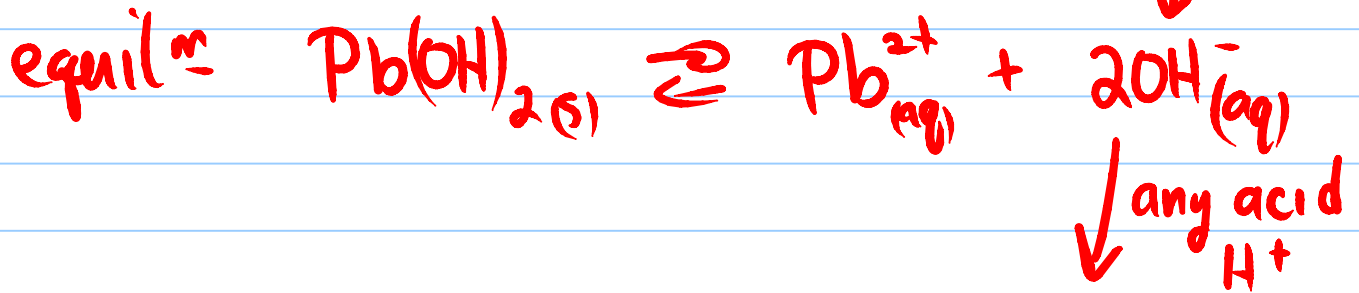
Increasing Solubility of Hydroxide Compounds

Note Title

4/16/2007

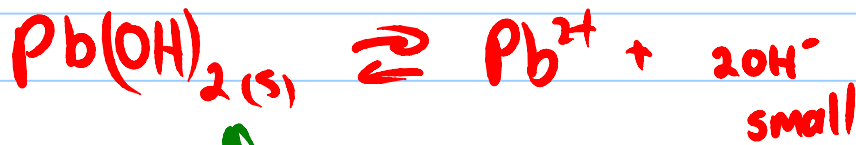
eg) $Pb(OH)_2$ low solubility

decreases $[OH^-]$



H_2O

shifts right →



↑
increase sol. of $Pb(OH)_2$